

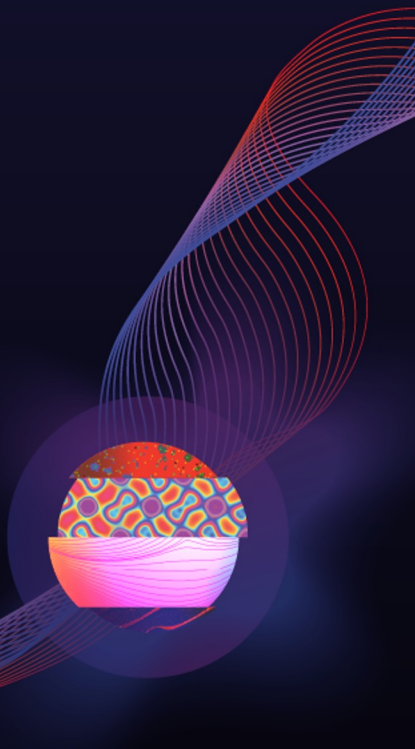
Exploring Ion Dynamics in Charged Systems using Real-Space Pseudopotential Density Functional Theory

Kai-Hsin Liou and James R. Chelikowsky

McKetta Department of Chemical Engineering, University of Texas at Austin

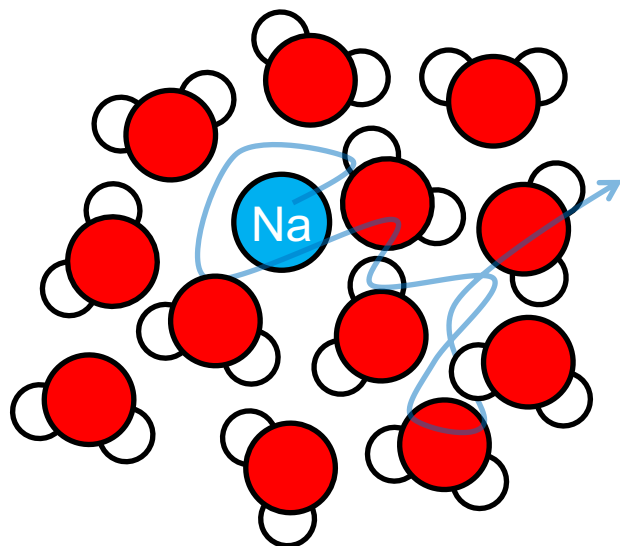
March 15, 2022

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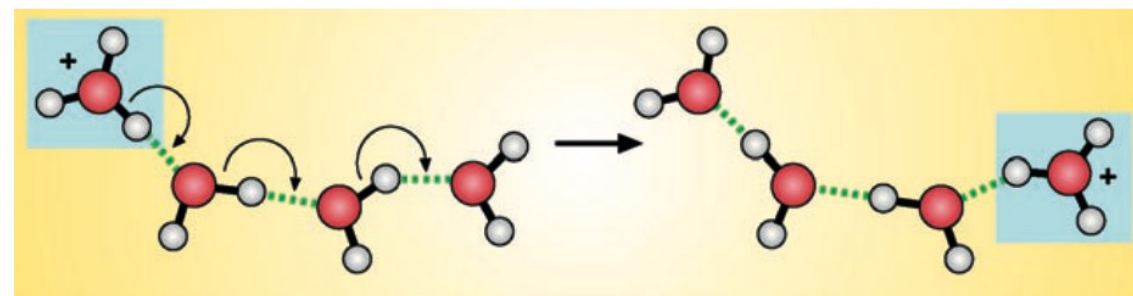


Proton transfer in water

Motion of sodium ions in water



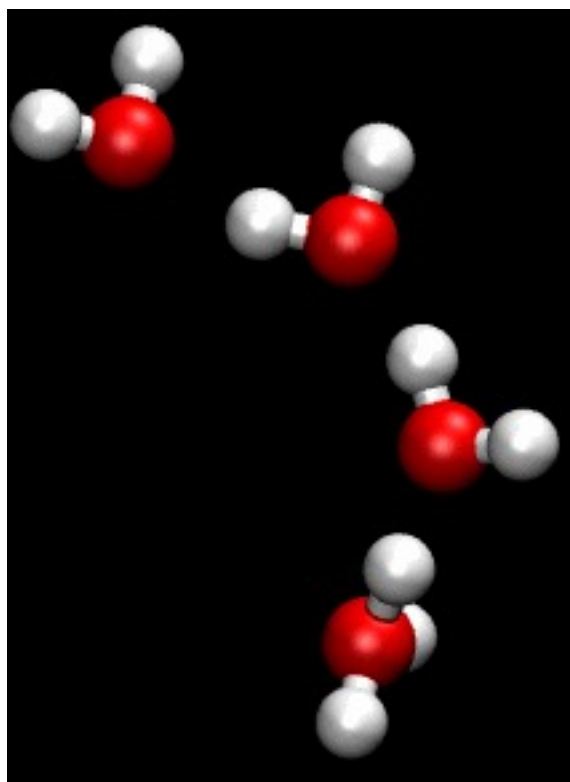
Motion of protons in water
(Grotthuss mechanism)



Marx, D., *ChemPhysChem* **2006**, 7 (9), 1848-70.



Proton transfer in water



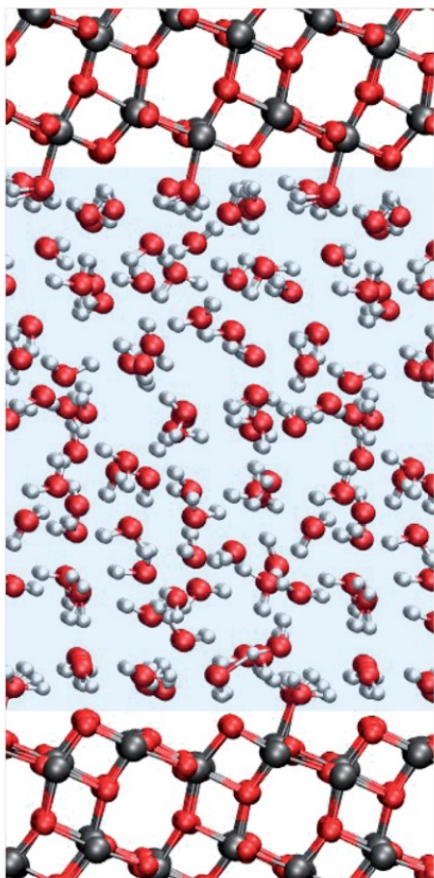
Wikipedia “Grotthuss mechanism”

Diffusivity of common ions in aqueous solution at 298 K

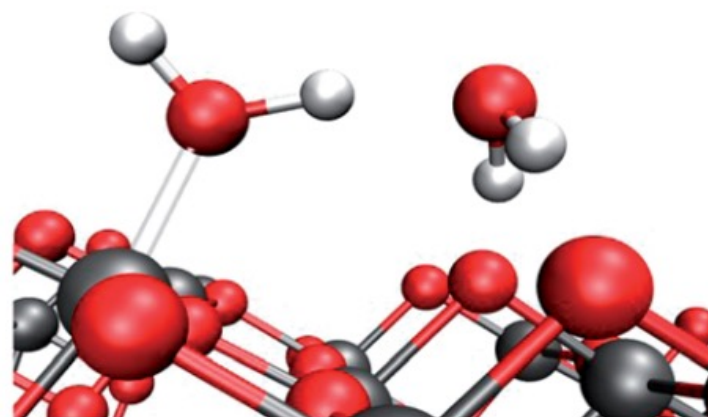
Ion	D (ang ² ps ⁻¹)
H ⁺	0.9311
D ⁺	0.6655
Li ⁺	0.1029
Na ⁺	0.1334
K ⁺	0.1957
NH ₄ ⁺	0.1957

Haynes, W. M., *CRC Handbook of Chemistry and Physics*. CRC Press: 2016.

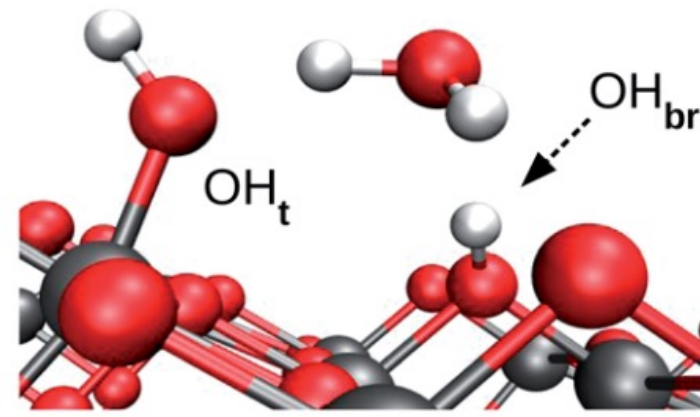
Interfaces between TiO_2 and liquid water



Molecular adsorption



Dissociative adsorption

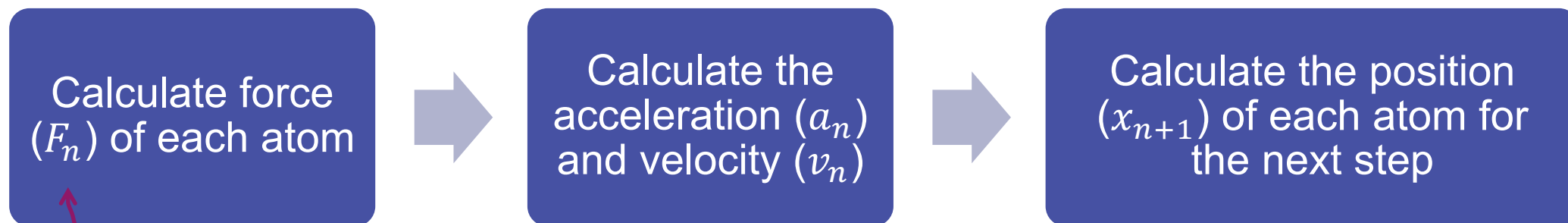




ab-initio molecular dynamics

at step n

to step $n + 1$



From first-principles calculations



Real-space pseudopotential density functional theory

The Kohn–Sham equations:

$$\left[-\frac{1}{2} \nabla^2 + \hat{V}_{\text{ion}} + V_{\text{H}}(\mathbf{r}) + V_{\text{xc}}(\mathbf{r}) \right] \phi_i(\mathbf{r}) = \varepsilon_i \phi_i(\mathbf{r})$$

$\hat{V}_{\text{ion}}$ ionic pseudopotential

$V_{\text{H}}(\mathbf{r})$ Hartree potential

$V_{\text{xc}}(\mathbf{r})$ exchange-correlation potential



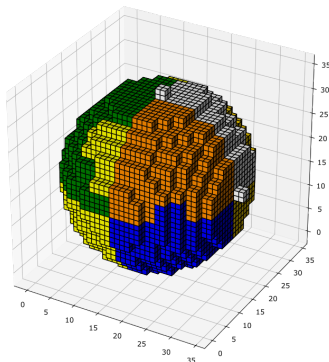
Chebyshev-filtered subspace iteration method

$$\overbrace{\left[-\frac{1}{2} \nabla^2 + V_{\text{eff}} \right]}^{H_{\text{KS}}} \phi_i = \varepsilon_i \phi_i$$

$$\{\phi_i\}_{i=1,\dots,N_S}$$

The subspace of the lowest-energy states

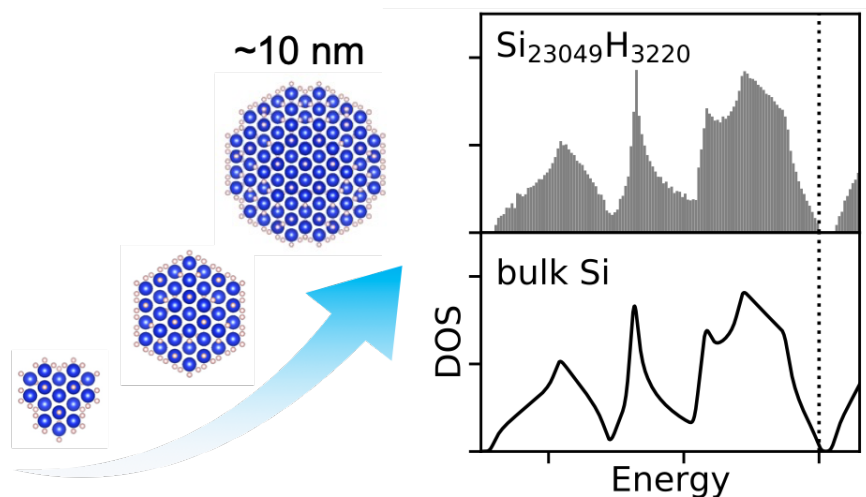
1. **Filtering** the subspace to amplify the lowest-energy states
2. **Orthonormalizing** the subspace
3. Obtaining the approximate eigenpairs from the subspace via **Rayleigh–Ritz method**



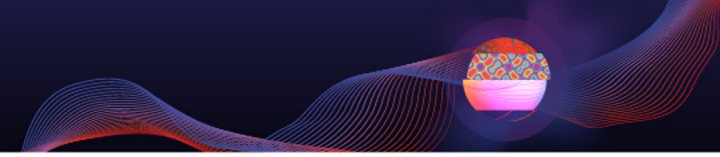
PARSEC

<http://real-space.org/>

Pseudopotential Algorithm for Real-Space Electronic Structure Calculation



System	# of grid points	# of states	# of Cori KNL nodes	Wall time (hrs)
Si ₈₄₉ H ₃₄₈	812,112	2,000	1	1.3
Si ₄₀₀₁ H ₁₀₁₂	2,707,504	9,216	8	4.0
Si ₁₀₈₆₉ H ₁₉₂₄	6,377,184	24,576	64	8.4
Si ₂₃₀₄₉ H ₃₂₂₀	15,180,904	61,440	256	27.9



Setup for liquid water simulations



Number of H ₂ O	64
Density (g cm ⁻³)	0.996
Grid spacing (bohr)	0.25 (~158 Ry of kinetic energy cutoff)
Time step (fs)	0.5
XC functional	GGA-PBE
Pseudopotential	Troullier–Martins norm-conserving
Coefficient of the Langevin thermostat, β	0.01



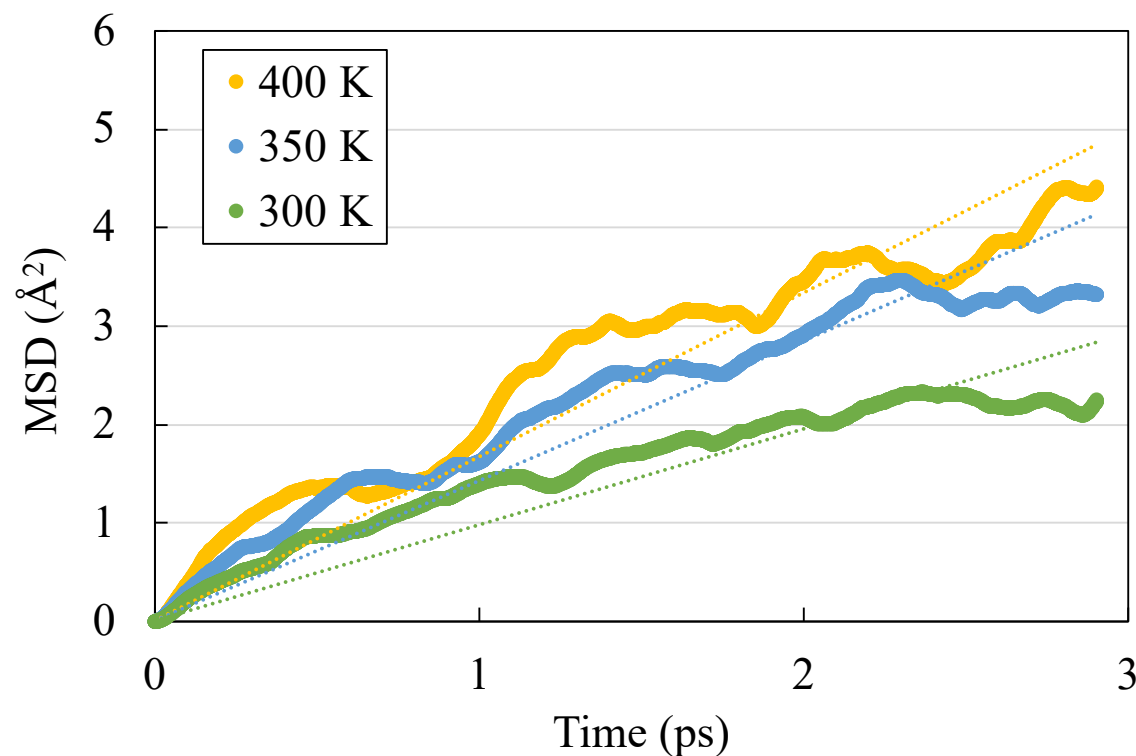
Diffusivity of H₂O in liquid water

Simulations

Temperature (K)	D (ang ² ps ⁻¹)
400	0.278
350	0.237
300	0.163

Experiments

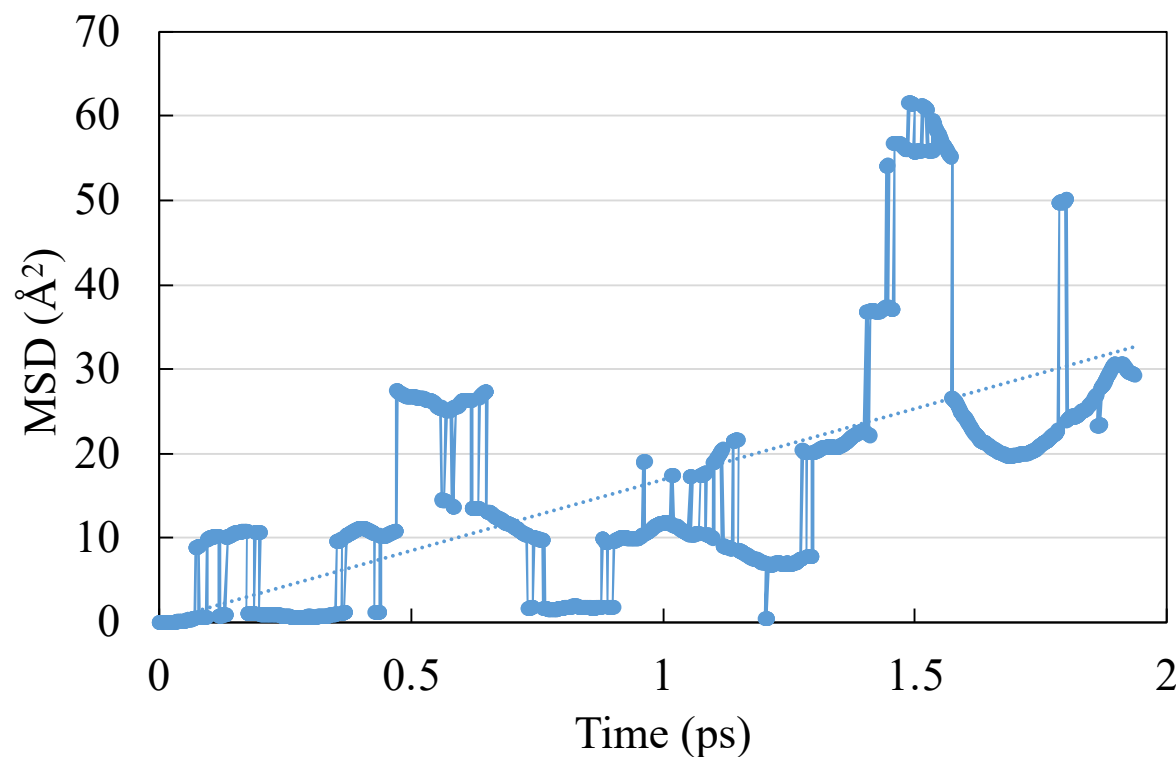
Temperature (K)	D (ang ² ps ⁻¹)
318	0.3575
308	0.2919
298	0.2299
288	0.1777
278	0.1313



Mills, R., *The Journal of Physical Chemistry* **1973**, 77 (5), 685-688.



Diffusivity of H^+ in liquid water



Simulations

Temperature (K)	D ($\text{\AA}^2 \text{ps}^{-1}$)	$D_{\text{H}^+}/D_{\text{H}_2\text{O}}$
440	3.00 [1]	5.4 [1]
350	2.69 ± 1.33	11.4
300	1.02 [1]	23.1 [1]

Experiments

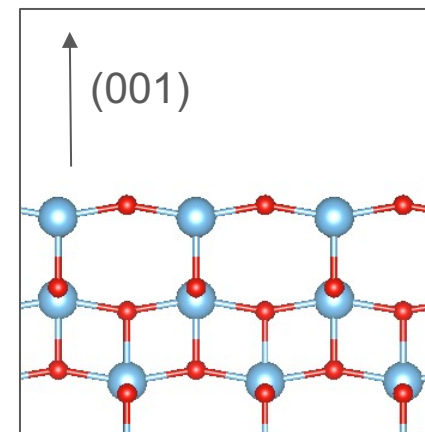
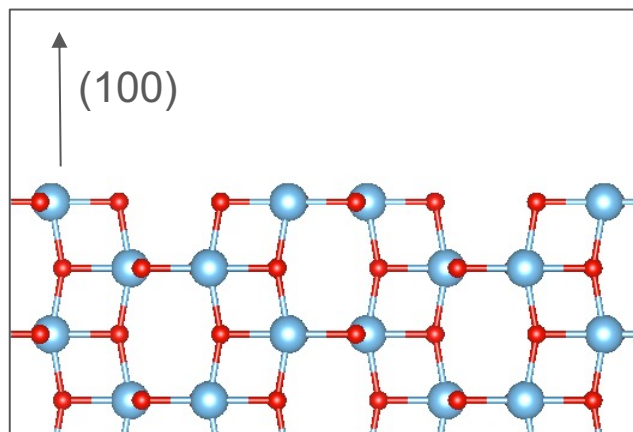
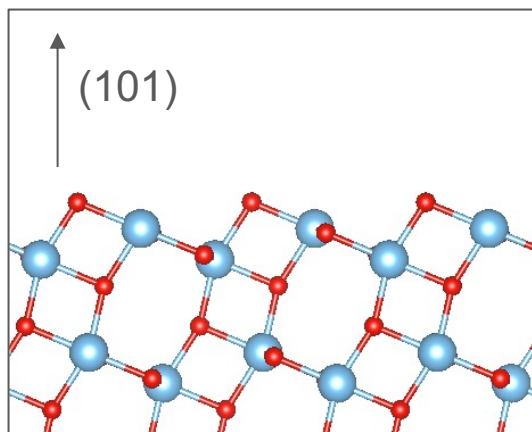
Temperature (K)	D ($\text{\AA}^2 \text{ps}^{-1}$)	$D_{\text{H}^+}/D_{\text{H}_2\text{O}}$
298	0.9311 [2]	4.05 [3]

[1] Fischer, S. A.; Dunlap, B. I.; Gunlycke, D., *Chemical Science* **2018**, 9 (35), 7126-7132.

[2] Haynes, W. M., *CRC Handbook of Chemistry and Physics*. CRC Press: 2016.

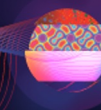
[3] R. Mills and V. M. M. Lobo, Elsevier, New York, NY, 1989, vol. 36.

Surface energy of anatase TiO_2



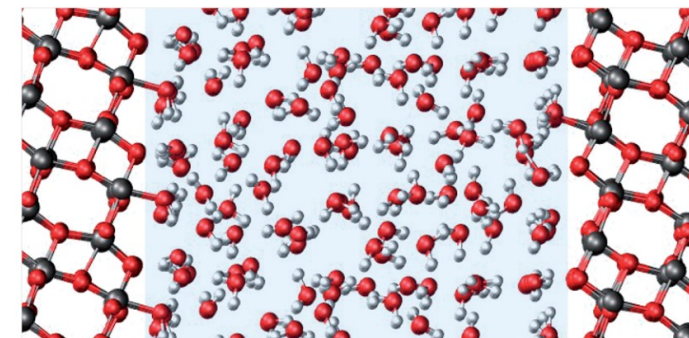
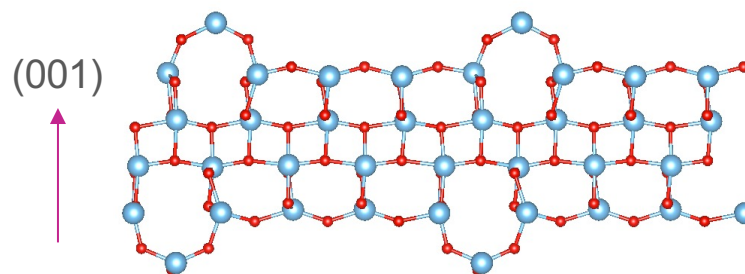
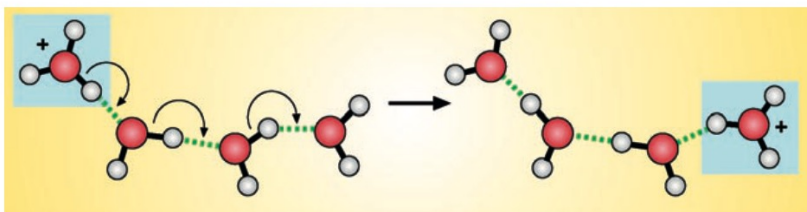
Surface	Surface energy (J/m ²)	
	PARSEC	Literature[1]
(101)	0.55	0.49
(100)	0.59	0.58
(001)	1.10	0.98

[1] Lazzeri, M.; Vittadini, A.; Selloni, A., *Physical Review B* **2001**, 63 (15), 155409.

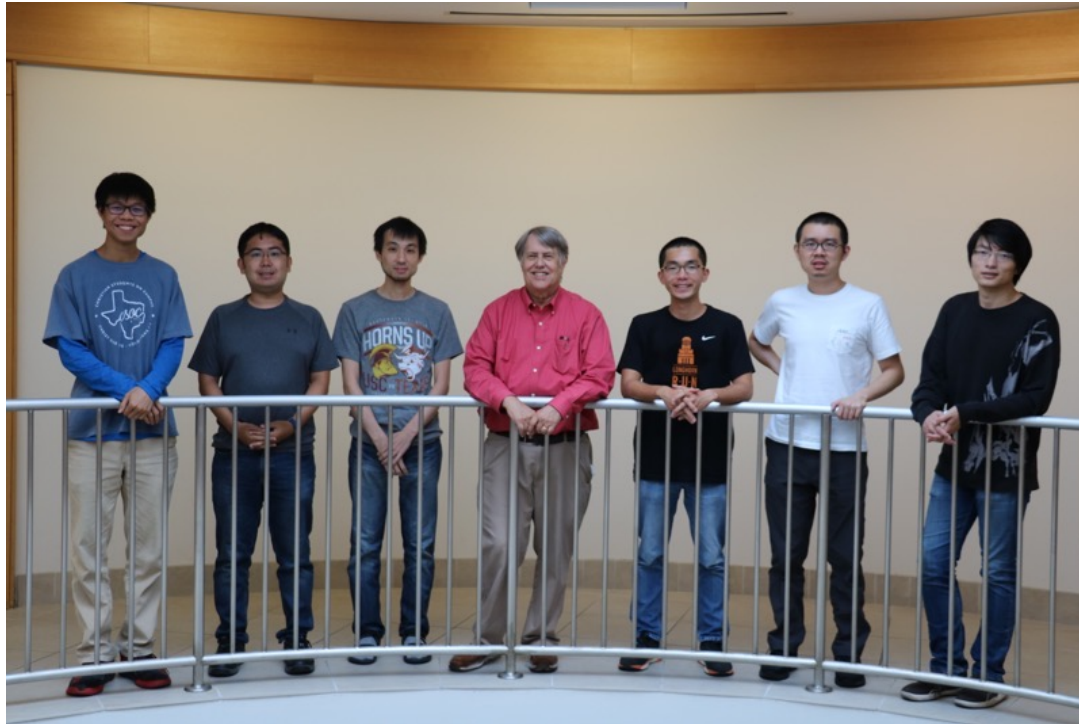


Summary and future work

- *ab-initio* MD using real-space pseudopotential density functional theory
- Proton transfer in liquid water
- Reconstruction and energetics of TiO_2 surfaces
- Dynamics at TiO_2 –liquid water interfaces and their optical responses



Acknowledgements



Center for Computational Materials



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